



# Development of HPLC Determination Method for Trace Levels of 1-, 2-Nitropyrenes and 2-Nitrofluoranthene in Airborne Particulates and Its Application to Samples Collected at Noto Peninsula

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## ABSTRACT

1-Nitropyrene (1-NP), 2-NP and 2-nitrofluoranthene (2-NFR) are useful markers for studying the atmospheric behaviors of polycyclic aromatic hydrocarbons (PAHs) and nitropolycyclic aromatic hydrocarbons (NPAHs). However, present methods for measuring trace levels of these compounds are less-sensitive and laborious. Here we describe several improvements to a previously reported high-performance liquid chromatography-chemiluminescence detection system that allows it to determine trace levels of 1-, 2-NPs and 2-NFR. The proposed system was equipped with a reducer column packed with Pt/Rh instead of zinc whose life-time was limited. The combination of Cosmosil MS-II (monomeric ODS) and AR-II (polymeric ODS) columns was used instead of polymeric ODS columns as the separator column to improve the separation. An ethanol mixture with acetate buffer (pH 5.5) was used in place of an acetonitrile mixture with the same buffer to activate the reducer column. The same ethanol mixture was used as the mobile phase for the clean-up column. The switching time of the column switching valve was optimized to concentrate the amino-derivatives of above NPAHs quantitatively on the concentrator column. The concentrations of bis(2,4,6-trichlorophenyl) oxalate and hydrogen peroxide in the chemiluminescence reagent solution were optimized to 0.4 mM and 30 mM, respectively, to increase the sensitivity. Under the above conditions, the detection limits ( $S/N=3$ ) of 1-, 2-NPs and 2-NFR were 1 fmol (0.25 pg), 10 fmol (2.5 pg) and 4 fmol (1 pg), respectively. The proposed system was effectively used to determine trace levels of 1-, 2-NPs and 2-NFR in airborne particulates collected at Noto Peninsula. The atmospheric concentrations of 1-, 2-NPs and 2-NFR were not more than  $\text{sub } \mu\text{g m}^{-3}$  levels. They were higher in

winter (January) than in summer (July). In both seasons, the concentrations were in decreasing order,  $[2\text{-NFR}] > [1\text{-NP}] > [2\text{-NP}]$ .

**Key words:** 1-Nitropyrene, 2-Nitropyrene, 2-Nitrofluoranthene, Trace analysis, Airborne particulate matter

## 1. INTRODUCTION

Among the air pollutants exhausted from the combustion of fossil fuels such as petroleum and coal, polycyclic aromatic hydrocarbons (PAHs) show carcinogenicity and/or mutagenicity (Tokiwawa *et al.*, 1980), endocrine disrupting activity (Kizu *et al.*, 2000) or reactive oxygen species producing activity (Motoyama *et al.*, 2009). Nitropolycyclic aromatic hydrocarbons (NPAHs) such as 1-nitropyrene (1-NP) and 1,3-dinitropyrene are also exhausted from the combustion of fossil fuels, and their mutagenicity was much stronger than that of benzo[*a*]pyrene. Additionally, several NPAHs such as 2-NP and 2-nitrofluoranthene (2-NFR) are subsequently formed by the reaction of PAHs and NO<sub>x</sub> in the air (Arey *et al.*, 1986), suggesting that 1-, 2-NPs and 2-NFR are markers for studying the atmospheric behaviors of PAHs and NPAHs. Therefore, a method for simultaneously determining these three NPAHs would be useful. The authors developed a highly sensitive determination method for NPAHs in environmental samples such as airborne particulates and particulates exhausted from automobiles by using high-performance liquid chromatography (HPLC) with chemiluminescence detection (CLD) (Hayakawa *et al.*, 1995, 1991; Imaizumi *et al.*, 1990). Although NPAH concentrations in urban air were high enough to be determined by HPLC-CLD, the concentrations were much lower in the air at suburban and background sites that were not

near any contributors such as factories or high-traffic roads. We collected one-week airborne particulates with a high-volume air sampler at Kanazawa University Wajima Air Monitoring Station in the Noto Peninsula, Japan, which was a background monitoring site. The data showed a seasonal variation of PAH concentrations caused by long-range transport from China (Yang *et al.*, 2007). However, no NPAH were detected.

We previously determined trace levels of 1-, 2- and 4-NPs, 2-NFR and 6-nitrochrysene (6-NC) in airborne particulates and precipitation collected in urban area by introducing reducer and concentrator columns into our conventional HPLC-CLD system (Murahashi and Hayakawa, 1997). However, the life time of the reducer column packed with zinc was limited and the analysis time took more than 90 minutes since three sample injections were necessary to analyze the NPAHs. Recently, we developed an HPLC-CLD system for the simultaneous determination of more than 20 NPAHs by introducing the reducer column packed with Pt/Rh instead of zinc, since Pt/Rh catalyzed the reduction of NPAHs. In the system, an ethanol-aqueous solution containing ascorbic acid was used to activate the reducer column instead of an acetonitrile-aqueous solution (Tang *et al.*, 2005, 2003; Hayakawa *et al.*, 2001). However, the resolution and sensitivity were not high enough to detect trace levels of above NPAHs.

In this report, we improved an HPLC-CLD system with lower detection limits and higher resolutions of 1-, 2-NPs and 2-NFR by improving our previous system (Murahashi and Hayakawa, 1997). The proposed method was used for determining trace levels of 1-, 2-NPs and 2-NFR in airborne particulate samples collected in winter and summer at Noto Peninsula.

## 2. MATERIALS AND METHODS

### 2.1 Chemicals

1-, 4-NPs and 2-NFR were purchased from Aldrich Company Inc. (Milwaukee, WI, U. S. A.). 2-NP was kindly supplied by Professor Akihisa Hirayama, Kyoto Pharmaceutical University. Deuterated 1-NP (1-NP- $d_9$ ) was purchased from CDN Isotopes Inc. (Pointe-Claire, Quebec, Canada). All other chemicals used were of analytical reagent grade.

### 2.2 HPLC System and Conditions

Three on-line reducers, including an electrochemical reactor and columns packed with metals have been reported for reducing NPAHs to their corresponding amino-derivatives (Hayakawa *et al.*, 2001, 1993; Imaizumi *et al.*, 1990). The electrochemical reducer showed the lowest reduction efficiency and the other two showed quantitative reduction efficiency. The life time of

the zinc reducer column was limited, while the life time of the Pt/Rh reducer column was not (Hayakawa *et al.*, 2001, 1993). Considering these advantages of the Pt/Rh reducer column, we set up the HPLC-CLD system with a Pt/Rh reducer column according to our previous report (Tang *et al.*, 2003) with several modifications (Fig. 1). The system consisted of four HPLC mobile phase pumps, a chemiluminescence reagent solution pump, a chemiluminescence detector, a system controller, an integrator, a degasser, two column ovens for the reducer column (80°C) and for the clean-up and separator columns (20°C), an auto sample injector (100  $\mu$ L), a switching valve, a guard column, a concentrator column, a clean-up column, a reducer column (4.0 i.d.  $\times$  10 mm, packed with Pt/Rh, Shimadzu) and two separator columns (Cosmosil 5C18-MS-II, 4.6 i.d.  $\times$  150 mm, Nacalai Tesque and 5C18-AR-II, 4.6 i.d.  $\times$  250 mm, Nacalai Tesque) connected in series.

The mobile phases for the clean-up column and reducer column was ethanol-acetate buffer (pH 5.5) (3:1, v/v) at a flow rate of 0.3 mL min<sup>-1</sup>. The mobile phase for the concentrator column was the same ethanol-acetate buffer with 30 mM ascorbic acid at a flow rate of 1.8 mL min<sup>-1</sup>. The mobile phase for the separator columns was a mixture of acetonitrile and 10 mM imidazole buffer (pH 7.6) (1:1, v/v) at a flow rate of 1.0 mL min<sup>-1</sup>. The chemiluminescence reagent solutions were acetonitrile solutions containing 0.4 mM bis(2,4,6-trichlorophenyl)oxalate (TCPO) and 30 mM hydrogen peroxide, respectively, each cooled with ice-water. Both solutions were mixed (1:1, v/v) in the system, and the flow rate of the mixture was 1 mL min<sup>-1</sup>. Other conditions were the same as in our previous reports (Tang *et al.*, 2005, 2003).

### 2.3 Preparation of Standard Solutions

Stock standard solutions of 1-, 2-, 4-NPs, 2-NFR and 1-NP- $d_9$  were independently prepared by dissolving each crystal in ethanol. The stock standard solutions were mixed, and then the mixture was diluted adequately with acetonitrile-water (3:1, v/v) to give calibration curve solutions: 0.1-100 pmol L<sup>-1</sup> for 2-NFR; 0.4-400 pmol L<sup>-1</sup> for 2-NP; 0.05-50 pmol L<sup>-1</sup> for 1-NP. 1-NP- $d_9$  standard solution was at the concentration of 0.2 pmol L<sup>-1</sup> in ethanol.

### 2.4 Sampling and Sample Preparations

Airborne particulate samples were collected at Kanazawa University Wajima Air Monitoring Station at Noto Peninsula (Nisifutamata-machi, Wajima City, Ishikawa Prefecture, Japan). Airborne particulates were collected by a high volume air sampler (AH-600, Shibata Japan) with a quartz fiber filter (8  $\times$  10 inch, 2500QAT-UP, Pallflex Products, Putnam, CT, U. S. A.) at a flow rate of 0.7 m<sup>3</sup> min<sup>-1</sup> in January and July, 2007.

The filter was newly changed every week. Airborne particulates were collected from January 4 to 11, 2008, too. One-fourth of the filter (corresponding to  $3.53 \times 10^3 \text{ m}^3$  air) was cut into small pieces in a flask and 100  $\mu\text{L}$  of 1-NP- $d_9$  standard solution was added to the flask as an internal standard. Then, NPAHs were extracted ultrasonically twice with benzene/ethanol (3:1, v/v). The solution was washed successively with sodium hydroxide, sulfuric acid solution and water. After filtering the organic solution with an HLC-DISK membrane (pore size 0.45  $\mu\text{m}$ , Kanto Chemical Co., Tokyo, Japan), the solution was evaporated to dryness. The residue was dissolved in 1.0 mL of ethanol, and then an aliquot (100  $\mu\text{L}$ ) of the solution was injected into the HPLC-CLD system.

### 3. RESULTS AND DISCUSSION

#### 3.1 Comparison of Separator Columns

1-NP is one of the major NPAHs exhausted directly from combustion processes of fossil fuel, and both 2-NFR and 2-NP are major NPAHs which are secondarily formed in the air (Arey *et al.*, 1986). These facts

suggest that the simultaneous determination of 1-, 2-NPs and 2-NFR is useful for determining their atmospheric behaviors. In our previous HPLC-CLD system using the reducer column packed with Pt/Rh, however, the retention time of 2-NFR was close to that of 2-NP (Tang *et al.*, 2005). Moreover, the retention time of 4-NP was similar to the retention times of 2-NFR and 2-NP, and the concentration of 4-NP was much lower (Murahashi *et al.*, 1999). These observations suggest that the resolution of the NPAHs should be improved, resulting in more accurate determination.

Polymeric ODS columns are known to give better resolution of PAHs than monomeric ODS columns. However, the difference of those two-type ODS columns in the separation of amino-derivatives of PAHs (APAHs) has not been examined. We compared the retentions of amino-derivatives of 1-, 2-, 4-NPs (1-, 2-, 4-aminopyrenes, 1-, 2-, 4-APs) and 2-NFR (2-amino-fluoranthene, 2-AFR) on Cosmosil MS-II (monomeric ODS) and AR-II (polymeric ODS) columns by introducing them into the HPLC-CLD system as shown in Fig. 1 without the two guard columns and the clean-up column. A mixture of acetonitrile and aqueous buffer is suitable for the peroxyoxalate chemiluminescence

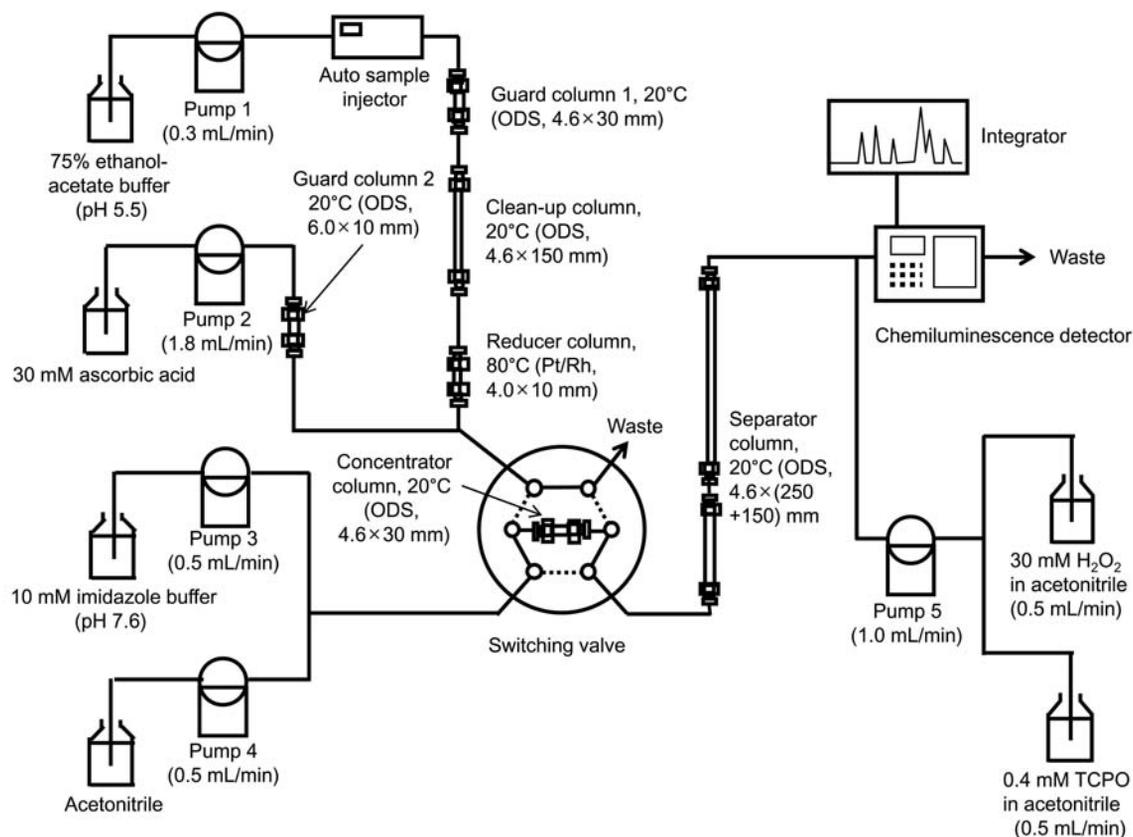


Fig. 1. Schematic diagram of the proposed HPLC-CLD system.

**Table 1.** Retention times (minute) of 1-, 2-, 4-NPs and 2-NFR<sup>1)</sup>.

Compound	MS-II (400 mm) <sup>2)</sup>	AR-II (400 mm) <sup>2)</sup>	MS-II (150 mm)+AR-II (250 mm) <sup>2)</sup>
2-NFR	48.5	54.5	52.5
2-NP	48.5	58	54.5
4-NP	51	58	57
1-NP	55.5	63.5	60.5

<sup>1)</sup>1-, 2-, 4-NPs and 2-NFR were reduced to their corresponding amino-derivatives in the reducer column and then introduced into the separator columns.

<sup>2)</sup>Inner diameter of the columns was 4.6 mm.

detection of amino-derivatives of PAHs (Hayakawa *et al.*, 2001, 1995). However, the reduction activity of Pt/Rh occurs in hot ethanol but not in acetonitrile, suggesting that the carrier solution going into the reducer column must contain ethanol but not acetonitrile. Therefore, a mixture of ethanol and acetate buffer (pH 5.5) was used as the mobile phase for the reducer column. On the other hand, amino-derivatives of PAHs were unstable in the system. Ascorbic acid was added to the effluent from the reducer column to keep the amino-derivatives stable in our previous HPLC-CLD system (Murahashi and Hayakawa, 1997). Thus, ascorbic acid was added to the mobile phase for the concentrator column in the proposed system as described in Experimental.

When NPAH standard solutions were injected into the system, NPAHs were all reduced quantitatively to their corresponding amino-derivatives in the reducer column. However, the separation of 2-AFR and 2-AP was impossible on a Cosmosil MS-II column, although the separation of 1- and 2-APs was complete. On the other hand, a Cosmosil AR-II column completely separated 2-AFR and 2-AP but did not completely separate 2- and 4-APs. By using a combination of Cosmosil MS-II and AR-II columns, the four APAHs were separately determined as amino derivatives (Table 1). The mean concentration ratio of 4-NP (0.57 fmol m<sup>-3</sup>) was around 1/100 or 1/6 of those of 2-NFR and 2-NP, respectively, in the urban air of Kanazawa (Murahashi and Hayakawa, 1997), suggesting that 4-NP is not a good primary or secondary formation marker such as 1-, 2-NPs and 2-NFR. Accordingly, 1-, 2-NPs and 2-NFR were selected as target compounds in the following experiments and the combination of Cosmosil MS-II column (4.6 i.d. × 150 mm)+Cosmosil AR-II column (4.6 i.d. × 250 mm) was used for the following experiments.

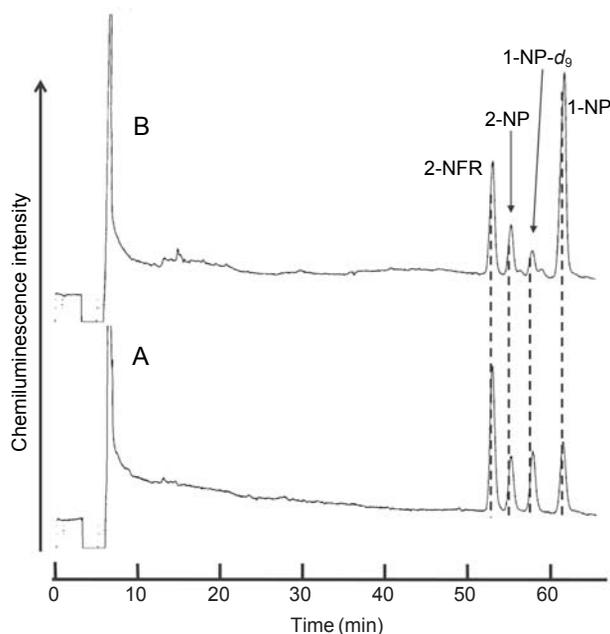
### 3.2 Optimum Composition of Chemiluminescence Reagent Solution

Next, we examined the optimal time for switching the switching valve after loading APAHs on the concentrator column of the HPLC-CLD system as shown in Fig. 1. When the standard mixture of 1-, 2-NPs, 2-

NFR and 1-NP-*d*<sub>9</sub> was injected into the system, they were eluted from the reducer column in the period from 17 to 29 minutes with the following elution order; 1-NP-*d*<sub>9</sub>=1-NP < 2-NFR < 2-NP. From this result, the concentration time (switching time of the switching valve) was set at 17-29 minutes.

The optimum composition of chemiluminescence reagent solution is known to be different for different analytes, and the chemiluminescence intensity is known to be affected by the concentration of organic solvent in the mobile phase (Hayakawa *et al.*, 1991). In view of the fact that the optimum acetonitrile concentration in the mobile phase was 50% for the elution and separation of 1-, 2-APs and 2-AFR, we examined the optimum concentrations of TCPO and hydrogen peroxide in the chemiluminescence reagent solution. The signal to noise (S/N) ratio of 1-AP increased with increasing TCPO concentration at concentrations above 0.02 mM and peaked at 0.15 mM TCPO. Furthermore, the S/N ratios of 2-AFR and 2-AP increased with increasing TCPO concentration, peaking at 0.4 mM TCPO. The S/N ratios of 2-AFR and 1-AP increased with increasing hydrogen peroxide concentration at concentrations above 4 mM and peaked at 15 mM hydrogen peroxide. On the other hand, the S/N ratios of 2-AP increased with increasing hydrogen peroxide concentration, peaking at 30 mM hydrogen peroxide. In view of the fact that the concentration of 2-NP in airborne particulates is usually lower than the concentrations of 2-NFR and 1-NP, the concentrations of chemiluminescence reagents should be optimum to obtain the highest sensitivity of 2-AP. Thus, we used 0.4 mM TCPO and 30 mM hydrogen peroxide in the following experiments.

Under the above conditions, the detection limits (S/N = 3) of 2-NFR, 2-NP and 1-NP by the proposed HPLC-CLD system were 4, 10 and 1 fmol, respectively. These values were better than the values obtained by the previous system. For example, the previous detection limit of 1-NP was 10 fmol (Tang *et al.*, 2005). The calibration curves were straight in the ranges from 10 fmol to 10 pmol (2-NFR), from 50 fmol to 50 pmol (2-NP) and from 5 fmol to 10 pmol (1-NP), respectively, with correlation coefficients over 0.999 and RSDs less than 5%.



**Fig. 2.** Typical chromatograms of (A) standard mixture of 1-, 2-NPs, 2-NFR and 1-NP-*d*<sub>9</sub> and (B) extracts from airborne particulates collected at the Noto Peninsula. (A) Airborne particulates were collected by a high-volume air sampler at a flow rate of 1,500 L min<sup>-1</sup> from January 4 to 11, 2008. One-fourth of the filter was used. (B) Injected amounts of 1-, 2-NPs, 2-NFR and 1-NP-*d*<sub>9</sub> were 5, 50, 50 and 5 fmol, respectively.

### 3.3 Determination of 1-, 2-NPs and 2-NFR Extracted from Airborne Particulates

Fig. 2 shows typical chromatograms of (A) the standard mixture of 1-, 2-NPs and 2-NFR and (B) the extracts from airborne particulates collected at Kanazawa University Wajima Air Monitoring Station on the Noto Peninsula, Japan from January 4 to 11, 2008. 2-NFR, 2-NP and 1-NP were chemilumigenically determined as their corresponding amino-derivatives in chromatogram B. This result suggests that the proposed HPLC-CLD system is suitable for the determination of NPAHs in air samples collected at background monitoring sites. The calculated concentrations of 1-NP, 2-NP and 2-NFR were 1.8 fmol (450 fg) m<sup>-3</sup>, 0.65 fmol (160 fg) m<sup>-3</sup> and 6.0 fmol (1,500 fg) m<sup>-3</sup>, respectively. The stable baseline of chromatogram A suggests that the proposed system can determine these NPAHs at even much lower concentration.

### 3.4 Atmospheric Concentrations of 1- and 2-NPs and 2-NFR at Noto Peninsula

1-, 2-NPs and 2-NFR in airborne particulates collected in January and July, 2007 at Noto Peninsula were determined by the proposed method. The atmospheric concentrations were not more than sub pg m<sup>-3</sup> levels.

**Table 2.** Atmospheric concentrations (fg m<sup>-3</sup>)<sup>1)</sup> of 1-, 2-NPs and 2-NFR at Noto Peninsula.

Month <sup>2)</sup>	1-NP	2-NP	2-NFR
January	840 ± 280	230 ± 100	1900 ± 520
July	80 ± 33	56 ± 31	510 ± 290

<sup>1)</sup>Each data means mean ± SD of four different week samples.

<sup>2)</sup>Samplings were performed in January and July, 2007. The filter was newly changed every week.

They were compared in Table 2. The concentrations of 1-, 2-NPs and 2-NFR were higher in winter (January) than in summer (July) by the factors of 1/3.7, 1/10.5 and 1/4.1, respectively. In both seasons, the concentrations were in decreasing order, [2-NFR] > [1-NP] > [2-NP]. It has been reported that PAHs exhausted from coal-burning systems especially in winter season in North-East China were long-range transported to Noto Peninsula over the Japan Sea (Yang *et al.*, 2007). The concentration level of 1-NP at Noto Peninsula was much lower than that at Shenyang (annual average concentration 107 ± 81 pg m<sup>-3</sup> (Hattori *et al.*, 2007)) by the factor of 3 or 4 orders of magnitude. This is the first report that the atmospheric concentrations of 1-, 2-NPs and 2-NFR at Noto Peninsula were determined. The proposed method is useful to know the behaviors of trace levels of NPAHs at background monitoring sites such as Noto Peninsula. The long-range transport of NPAHs from Asian Continent to Japan clarified by the proposed method will be reported elsewhere.

## 4. CONCLUSIONS

A HPLC-CLD system has been developed for the determination of trace levels of 1-, 2-NPs and 2-NFR in airborne particulates by modifying our previous system. The new system uses a reducer column packed with Pt/Rh, a chemiluminescence detector and a combination of monomeric and polymeric ODS columns for the separation column. The switching time of the switching valve and the concentrations of TCPO and hydrogen peroxide in the chemiluminescence reagent solution were optimized. The detection limits (S/N=3) of 1-, 2-NPs and 2-NFR by the proposed system were 1, 10 and 4 fmol, respectively. The proposed system determined trace levels of 1-, 2-NPs and 2-NFR in airborne particulates collected at Noto Peninsula. The atmospheric concentrations were not more than pg m<sup>-3</sup> levels.

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